١	The antimicrobial and antibiofilm potential of Citrus aurantium and Artemisia
۲	annua essential oils nanoemulsions
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#### **to** ABSTRACT

۲٦ Antimicrobial resistance has posed considerable health and economic burdens globally ۲۷ (approximately five million deaths annually), particularly among developing countries. The estimated annual treatment costs in the United States include US\$4.6 billion. Vast antibiotic ۲۸ ۲٩ resistance among Gram-negative and Gram-positive bacterial species has spread from healthcare ۳. to the environment, community, and animals. These conditions have limited and sometimes ۳١ failed the infection eradication choices and facilitated the distribution of drug-resistant organisms. The spread of drug-resistant bacterial infections is a huge human health concern, ٣٢ hence seeking novel antibacterial agents is crucial. This study used the nanoemulsions of Citrus ٣٣ aurantium and Artemisia annua essential oils (EOs) as natural antibacterial agents. Gas ٣٤ chromatography mass spectrometry (GC-MS) analysis showed that limonene (31.4%) and ۳0 37 artemisia ketone (26.2%) were major components, respectively. After that, their nanoemulsion dosage forms with a mean droplet size of  $181 \pm 7$  and  $160 \pm 5$  and with zeta potential values 3.1 ۳۷  $\pm$  0.8 and -4.9  $\pm$  0.5 mV were prepared. Meanwhile, successful loading of the EOs in ۳۸ ۳٩ nanoemulsion was confirmed by Attenuated Total Reflection-Fourier Transform Infrared (ATR-٤٠ FTIR) analysis. A. annua nanoemulsion with 40% antioxidant effect was significantly more potent than C. aurantium nanoemulsion. Meanwhile, nanoemulsions' antibacterial and ٤١ ٤٢ antibiofilm activity against clinical and standard strains, Escherichia coli, Staphylococcus ٤٣ aureus, Pseudomonas aeruginosa, and Klebsiella pneumonia, were investigated. The best ٤٤ efficiency was related to the effect of C. aurantium nanoemulsion against S. aureus; minimum 20 inhibitory and bactericidal concentrations (respectively MIC and MBC) were 500 and > 2000 ٤٦  $\mu$ g/mL. Besides, no biofilm was formed after treatment by both nanoemulsions. Therefore, C.

- *aurantium* and *A. annua* EO nanoemulsions may act as natural antioxidant and antibacterial
   agents in complementary medicine.
- **Keywords:** Antibacterial Agents, Essential Oil, Reactive Oxygen Species, Phytochemicals

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#### 1. Introduction

01 Vast antibiotic resistance among Gram-negative and Gram-positive bacterial species has ٥٢ spread from healthcare to the environment, community, and animals. These conditions have limited and sometimes failed the infection eradication choices, resulting in the distribution of ٥٣ drug-resistant organisms (1, 2) such as those non-susceptible to last-resort antibiotics of 0 2 00 carbapenems and glycopeptides. As alternatives to chemotherapies, natural resources such as essential oils (EOs) have been suitable alternatives (3-5). In addition, nanoformulation of EOs by ٥٦ optimization in particle size, increasing solubility, and improving bioavailability and stability ٥٧ causes medicines to be more permeable into cells, such as bacteria or cancer cells (6-10). The ٥٨ citrus genus includes several species of citrons within the family Rutaceae, utilized as herbal 09 ٦. medicines, fruits, juice, and additives. Citrus fruits contain vitamins C and B, minerals, nutrients, and bioactive compounds such as phenolic compounds, volatile oils, and terpenoids. Citrus ٦١ aurantium (C. aurantium) L. cultivar has exhibited anti-inflammatory, hypoglycemic, ٦٢ antimicrobial, anticancer, pain-relief, and organ protective effects (11). Although the species has ٦٣ ٦٤ bioactive compounds and biological activities, its pharmacological effects, traditional usage, and ٦0 exact bioactive compounds have not been uncovered (12). On the other, C. aurantium L. dietary supplementation has not exerted side effects (13). Various extracts of C. aurantium L. leaves, 77 ٦٧ including aqueous, alcoholic, or chloroform portions, have demonstrated antibacterial effects ٦٨ against various agents (11, 13). In addition, the antimicrobial effects of its EOs have unraveled ٦٩ potential activities against Agrobacterium tumefaciens, Dickeya solani, and Erwinia amylovora ٧. (14, 15). Artemisia annua grows globally in Europe, Asia and North America with preferred arid ۷١ and semi-arid climates and provides considerable health benefits such as anticancer and ۲۷ antimicrobial traits. Bioactive compounds of the herb mainly include artemisia ketone, 1,8-

۷۳	cineole, germacrene D, and camphor (16). This study aimed to investigate the antibacterial and
٧٤	antibiofilm effects of nanoemulsions of A. annua L. and C. aurantium.

#### 2. Materials and Methods

v٦ 2.1. Materials

C. aurantium and A. annua EOs were bought from Iranian companies, Tabib Daru ٧٧ ٧٨ Company and Pharmaceutical Company Essential Oil Dr. Soleimani. Bacterial standard strains ٧٩ included Escherichia coli ATCC25922, Staphylococcus aureus ATCC25923, Pseudomonas ٨. aeruginosa ATCC 27853, and Klebsiella pneumonia ATCC13883 were cultured from bacterial ۸١ stock preserved at -20 °C in the microbiology laboratory of Fasa University of Medical Sciences, ۸۲ Iran. Moreover, clinical isolates were collected from patients. Mueller Hinton broth, trypticase ۸۳ soy broth, crystal violet, tween 20, and tween 80 were purchased from Merck Chemicals, Germany. The standard analytical solution of Alkanes (C9-C24) was bought from Sigma-Aldrich ٨٤ ٨o (USA).

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#### **AV 2.2. Chemical composition of EOs**

Gas chromatography mass spectrometry (GC–MS) analysis was used for the chemical
 compositions of the EOs. For this purpose, the gas chromatographic device (Agilent 6890, USA)
 with HP-5MS silica fused columns coupled to a network mass selective detector (Agilent 5973,
 USA) was used. The major constituents of EOs were identified by comparing their retention
 indices to homologous C9-C24 n-alkanes, as described in our previous reports (17, 18).

# **97 2.3. Preparation and characterizations of nanoemulsions**

A. annua EO (0.4% v/v) was mixed with tween 80 (0.5% v/v), and *C. aurantium* (0.4%
 v/v) was mixed tween 20 (0.5% v/v), separately, for 3 minutes at room temperature at 2000 rpm.
 Afterward, distilled water was added drop by drop to reach the final volume of 5 mL, and the

٩٧ mixture was stirred for 40 minutes at room temperature at 2000 rpm. The prepared ٩٨ nanoemulsions' droplet size and droplet size distribution (SPAN) were measured using a DLS ٩٩ (Dynamic Light Scattering, DLS9900, K-ONE, Korea) device. SPAN was calculated using the 1 . . relationship d90-d10/d50; in this equation, d is the diameter, 90, 10, and 50 percent of particles with a size smaller than the values mentioned. The droplet size below 200 nm and SPAN below 1.1 1.1 one were necessary conditions to confirm the appropriate size characteristics. TEM (Transitional 1.7 Electron Microscopy, Philips, TEM, EM 208s, Netherland) was used to confirm the droplet size and determine their morphology. ATR-FTIR (Attenuated Total Reflection-Fourier Transform 1.2 1.0 Infrared) analysis is used to evaluate the successful loading of the EOs in the nanoemulsion. Spectra of the EOs, nanoemulsion (-oil), and nanoemulsion were recorded in 400 - 4000 cm<sup>-1</sup> 1.7 1.1 using a spectrometer (Tensor II model, Bruker Co, Germany). Furthermore, the stability of nanoemulsions was investigated. Nanoemulsions were centrifuged at -4, +4, and +25°C (14,000 1.4 g, 30 min) to investigate stability against precipitation. Besides, nanoemulsions were stored at 1.9 +45°C and room temperature for six consecutive intervals of 48 hours for thermal stability 11. analysis. Moreover, nanoemulsions were placed at -20°C and room temperature for six 111 117 consecutive 48-hour intervals for cryogenic stability. In addition, nanoemulsions were placed at  $4^{\circ}$  C and room temperature for six months for long-term stability analysis. After each test, the 117 115 nanoemulsion was visually checked for sedimentation, creaming, or biphasic.

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# 2.4. Investigation of antioxidant properties of EOs and nanoemulsions

DPPH (2,2-diphenyl-1-picrylhydrazyl) assay was used to measure antioxidant properties. First, serial dilutions of the nanoemulsions (62.5-2000  $\mu$ g/mL) were prepared in ethanol. Next, 50  $\mu$ L/well of each prepared dilution and 0.2 mM DPPH solution was added to a 96-well plate, and it was incubated for 30 minutes away from light at room temperature. Finally, the wells' OD (optical density) was read at 517 nm using a plate reader (Synergy HTX Multi-Model Reader,USA).

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The antioxidant activity was calculated using OD test/OD control×100.

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#### 2.5. Minimum inhibitory concentration

Micro-dilution test was implemented. The range of concentrations of nanoemulsions ١٢٤ 170 (250, 500, 1000, 1500 and 2000 µg/mL) was prepared by PBS containing 0.5% DMSO as 177 solvent. Antibacterial effects of EOs and nanoemulsion were investigated using 96-well broth micro-dilution, as described in our previous study (19). Briefly, 40 µL of each was inoculated ۱۲۷ ۱۲۸ into wells of 96 well plates containing 50  $\mu$ L of Mueller Hinton broth. Afterward, 10  $\mu$ L/well of each bacterial suspension (0.5 McFarland standard turbidity,  $1.5 \times 10^8$  CFU/mL) was added to 129 ۱۳. each well. The plates were incubated for 24 hours at 37 °C, and then the OD of the wells was read at 630 nm. Bacterial growth was obtained using OD sample/OD control×100. The minimum 171 ۱۳۲ inhibitory concentration (MIC) of A. annua and C. aurantium single and nanoemulsion forms ١٣٣ against bacterial strains were determined using concentrations ranging from 250-2000 µg/mL. Moreover, the bacterial suspension equal to 0.5 McFarland standard was prepared. The test was ١٣٤ 100 performed the same as that of the broth micro-dilution method.

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# 2.6.Biofilm formation

Anti-biofilm effects of EOs and nanoemulsions were assessed against clinical isolates. biofilmThe biofilm formation with and without exposure to the nanoemulsions was performed into 96-well plates using a microtitre tissue plate assay. Briefly, an overnight culture of bacterial strains was obtained into the trypticase soy broth (TSB) medium containing 1% glucose and diluted 1:100. For each bacterial suspension, 20  $\mu$ L was taken and inoculated into wells containing 180  $\mu$ L of the TSB medium in triplicate and incubated for 5 hours for exposed (2000 157 µg/mL of each nanoemulsion) group and 24h for the un-exposed group. The un-exposed group 122 medium was exchanged with each nanoemulsion and incubated for 24 hours. Next, the wells were washed using double distilled water and fixed using methanol. Then, 0.1% crystal violet or 120 127 safranin was added for 15 min. After repeated washing, the absolute ethanol (200  $\mu$ L) was added to solubilize bacterial contents and read using the ELISA reader at 490 nm. The biofilm ١٤٧ ١٤٨ formation levels (strong, moderate, weak, or non-adherence) were calculated using Table 1. This 129 study compared biofilm formation levels among groups, including the control (unexposed) and A. annua and C. aurantium EOs nanoemulsion-treated groups. 10.

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#### Table1. The calculation of biofilm formation levels

<b>Biofilm formation</b>	Calculation of cut-off	OD calculated results	Reference
ability	level		
Strong	OD>ODc×4	0.33296>OD	(20, 21)
Moderate	ODc×2≤OD <odc×4< td=""><td>0.16648≤OD&lt;0.33296</td><td></td></odc×4<>	0.16648≤OD<0.33296	
Weak	ODc≤OD<2×ODc	0.083324≤OD<0.16648	
No binding	OD≤ODc	0.08324≤OD	

OD: optical density, ODc: mean OD of control wells

#### **2.7.Data analysis**

All experiments were done in triplicates. The data was analyzed using the SPSS version, from which Chi-Square and analysis of variance (ANOVA) tests were applied to determine differences at a *p*-value cut-off of 0.05.

10V 3. Results

The main components of the *C. aurantium* were limonene (31.4%), sabinene (15.6%),  $\gamma$ terpinene (6.0%), linalool (5.6%) and cis-nerolidol (5.1%) (Table 2). As well as the main components of the *A. annua* were artemisia ketone (26.2%), camphor (19.2%), 1,8-cineole (12.3%), *trans*-caryophyllene (4.5%), and camphene (4.4%) (Table 2).

Retention	Compound	C. aurantium		A. annua		Retention
Time (min)		Area	%	Area	%	Index
9.46	α-pinene	58484827	1.7	122296938	4.1	932
10.06	camphene			132788539	4.4	954
11.14	sabinene	542668432	15.6	44039171	1.5	975
11.24	β-pinene	43301938	1.2	29556997	1.0	979
11.96	β-myrcene	108784770	3.1	29379228	1.0	988
12.50	Yomogi alcohol			41103264	1.4	999
13.08	a-terpinene	59177141	1.7			1014
13.81	1,8-cineole			368453758	12.3	1026
13.89	limonene	1088445097	31.4			1029
14.67	cis-ocimene	162728160	4.7	-	,	1037
15.13	y-terpinene	207125216	6.0			1054
15.48	artemisia ketone			784989266	26.2	1062
16.31	artemisia alcohol			33487881	1.1	1083
17.16	linalool	192637034	5.6			1095
19.22	camphor			576552736	19.2	1146
20.12	borneol			28483988	1.0	1169
20.63	4-terpineol	66692764	1.9	33586786	1.1	1177
23.55	cuminic aldehyde	128003231	3.7			1239
29.38	α-copaene			47639328	1.6	1376
31.26	trans-caryophyllene			136219018	4.5	1419
33.78	germacrene D			84811438	2.8	1481
33.99	β-selinene	/		90133377	3.0	1490
37.11	cis-nerolidol	178234586	5.1			1532
37.76	caryophyllene oxide			35663606	1.2	1583
42.89	cis-farnesol	34776356	1.0			1698

Table 2. Identified compounds in the EOs using GC-MS analysis

# **3.1. TEM analysis of** *C. aurantium* and *A. annua* EO nanoemulsions

The mean droplet diameter and zeta potential of *C. aurantium* nanoemulsion were  $181 \pm 7$  nm and  $3.1 \pm 0.8$  mV, respectively (**Figure 1 A & B**). The *A. annua* nanoemulsions showed droplet diameter at  $160 \pm 5$  nm and zeta potential values  $-4.9 \pm 0.5$  mV (**Figure 2 A & B**). The TEM analysis (**Figures 1C and 2C**) revealed that both nanoemulsions were spherical, size < 100 nm.

۱۷. Furthermore, the stability of nanoemulsions was investigated. Nanoemulsions were centrifuged at -4, +4, and +25°C (14,000 g, 30 min); no sedimentation or bi-phasic was 171 observed. Besides, nanoemulsions were stored at +45°C and room temperature for six 177 ۱۷۳ consecutive intervals of 48 hours for thermal stability analysis; no sedimentation or bi-phasic ١٧٤ was observed. Moreover, nanoemulsions were placed at  $-20^{\circ}$ C and room temperature for six consecutive 48-hour intervals for cryogenic stability; no sedimentation or bi-phasic was 140 177 observed. In addition, nanoemulsions were placed at 4 and room temperature for six months for long-term stability analysis; no sedimentation or bi-phasic was observed. 177







# 145 3.2. ATR-FTIR analysis of *C. aurantium* and *A. annua* nanoemulsions

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ATR-FTIR analysis confirmed EO loading in nanoemulsion (**Figure 3**). The spectra of *C*. *aurantium* EO displayed in Figure 3A, broadband at 3469 cm<sup>-1</sup> can be related to stretching vibration of the hydroxyl group due to hydrogen bonding in alcoholic and phenolic bioactive compounds in EO, spectra at 3076 cm<sup>-1</sup> can be corresponded to stretching vibration of CH in sp<sup>2</sup> groups and the bands at 2961, 2923 and, 2872 cm<sup>-1</sup> can be attributed to stretching vibration of

CH in sp<sup>3</sup> groups, the band at 1706 and 1676 cm<sup>-1</sup> can be related to stretching vibration of 19. 191 carbonyl groups. Bands at 1108 and 1052 cm<sup>-1</sup> showed stretching vibration of C-O groups. The peak at 989 cm<sup>-1</sup> is related to C-H bending absorption, and the strong peak at 758 cm<sup>-1</sup> is 198 allocated to benzene rings C-H vibration absorption. A peak at 689 cm<sup>-1</sup> is attributed to the ۱۹۳ vibration absorption of alkenes. The spectrum of nanoemulsion without C. aurantium EO 192 displayed in Figure 3B, the broad peak at about 3493 cm<sup>-1</sup> can be attributed to OH stretching 190 vibration due to hydrogen bonding between water and tween 20. Spectra at 2924 cm<sup>-1</sup> 197 corresponded to C-H stretching in tween 20. A strong band at 1733 cm<sup>-1</sup> attributed to C=O 197 stretching representing the carbonyl group in tween 20. Characteristic band at around 1462 cm<sup>-1</sup> ۱۹۸ can be related to CH<sub>2</sub> bending tween 20. Characteristic and sharp peak at 1091 cm<sup>-1</sup> is assigned 199 ۲.. to C-O stretching. FTIR of C. aurantium EO nanoemulsion spectrum (Figure 3C) showed the ۲.۱ broadband at 3518 cm<sup>-1</sup> attributed to OH stretching vibration due to the strong hydrogen bonding ۲.۲ between water, tween 20, and phenolic and alcoholic compounds in EO. Any band at 2969, 2924, and 2856 cm<sup>-1</sup> is related to C-H stretching due to sp<sup>3</sup> hybrid in tween 20 and EO. Strong ۲.۳ band at 1728 cm<sup>-1</sup> showed carbonyl stretching (C=O) tween 20 and EO. The absorption at around ۲.٤ 1456 cm<sup>-1</sup> corresponded to CH<sub>2</sub> bending tween and EO. A sharp and strong peak at about 1093 ۲.0 cm<sup>-1</sup> can be attributed to C-O stretching. ۲.٦

Spectrum of the *A. annua* EO has been demonstrated in **Figure 3D**, a broad and characteristic band at about 3520 cm<sup>-1</sup>, can be attributed to the hydroxyl functional groups in EO, and a band at 3084 cm<sup>-1</sup>, allocated to the stretching vibration of =C-H groups from olefins in sp<sup>2</sup> hybrid. Peaks at 2963, 2928, and 2872 cm<sup>-1</sup>, related to stretching vibrations of –CH in sp<sup>3</sup> hybrid, the spectra at around 1743 cm<sup>-1</sup> related to C=O, the absorption around 1620 and 1415 cm<sup>-1</sup> assigned to C = C, the bands at 1215 and 1167 cm<sup>-1</sup> are related to (C-O-C) bonds and the band at 876 cm<sup>-1</sup> can be

۳۱۲ allocated to angular deformations of CH<sub>2</sub> groups. Spectra of nanoemulsion without A. annua EO 212 has been shown in Figure 3E. A characteristic and broad peak between 3200 to 3600 cm<sup>-1</sup> 110 corresponds to OH stretching vibration due to the hydrogen bonding between water and tween 80. Spectra at 2964 and 2925 cm<sup>-1</sup> are related to C-H stretching. Absorption at 1740 cm<sup>-1</sup> was ۲۱٦ attributed to carbonyl stretching (C=O) in tween 80 and at about 1456 cm<sup>-1</sup> can be allocated to ۷۱۲ CH<sub>2</sub> bending. A strong and characteristic band at 1080 cm<sup>-1</sup> corresponded to C-O stretching. 212 219 ATR-FTIR of nanoemulsion containing A. annua EO has been displayed in Figure 3F. A peak at about 3422 cm<sup>-1</sup> is attributed to OH stretching vibration due to hydrogen bonding between EO, ۲۲. tween 80, and water and at 2924 cm<sup>-1</sup> corresponds to C-H stretching of EO and tween 80. ٢٢١ Absorption at 1733 cm<sup>-1</sup> attributed to C=O stretching representing the carbonyl group in EO and 222 tween. A characteristic peak at about 1462 cm<sup>-1</sup> outlined CH<sub>2</sub> bending in EO and tween 80. A ۲۲۳ strong and sharp peak at 1091 cm<sup>-1</sup> was assigned to C-O stretching. The presence of other bands 222 in EOs and blanks confirmed the successful loading of EOs in the prepared nanoemulsion. 220



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Figure 3. ATR-FTIR spectra of A: *C. aurantium* EO, B: nanoemulsion without *C. aurantium* EO, C:
 nanoemulsion containing *C. aurantium* EO, D: *A. annua* EO, E: nanoemulsion without *A. annua* EO, F:
 nanoemulsion containing *A. annua* EO

The *C. aurantium* and *A. annua* nanoemulsions were evaluated for their antioxidant effect by DPPH assay. As shown in **Figure 4**, the most potent free radical scavenging activity was obtained from *A. annua* EO nanoemulsion, 40 % at 2000  $\mu$ g/mL.





**Figure 4.** The antioxidant effects of samples

# YTA3.3. The minimum inhibitory concentration (MIC) and minimum bactericidalYTAconcentration (MBC)

From Table 3, the *C. aurantium* EO nanoemulsion MIC against *S. aureus, E. coli, P. aeruginosa,* and *K. pneumonia* included 500 µg/mL, 1000 µg/mL, 1000 µg/mL, and 1000 µg/mL, respectively. The MBC values against bacterial strains also included >2000µg/mL.
Moreover, The MIC values of *A. annua* nanoemulsion EO against *S. aureus, E. coli, P. aeruginosa,* and *K. pneumonia* included 1000 µg/mL, 2000 µg/mL, and 2000 µg/mL, respectively. The MBC values included >2000µg/mL, 2000 µg/mL, and 2000 µg/mL, respectively. The MBC values included >2000 µg/mL, 2000 µg/mL, and 2000 µg/mL, respectively. The MBC values included >2000 µg/mL for all the tested bacterial strains.

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# Table 3. The MIC and MBC levels (µg/mL) samples

Samples	S. aureus	E. coli	P. aeruginosa	K. pneumonia
	(MIC, MBC)	(MIC, MBC)	(MIC, MBC)	(MIC, MBC)
<i>C. aurantium</i> EO nanoemulsion	500, >2000	1000, >2000	1000, >2000	1000, >2000

A. annua EO	1000, >2000	2000, >2000	2000, >2000	2000, >2000
nanoemulsion				,

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#### ۲٤٨ **3.4.Antibacterial effects**

The bacterial growth inhibitory effect of *C. aurantium* nanoemulsion EO was concentration-dependent. The highest bactericidal effect was observed against *S. aureus* at 2000  $\mu$ g/mL, in which 56% of growth was inhibited (**Figure 5**).



# Figure 5. The bacterial growth in exposure to *C. aurantium* nanoemulsion EO The growth inhibitory effect of *A. annua* nanoemulsion EO was mostly against *S. aureus* at 2000 µg/mL, at which ~20 % of bacterial growth was inhibited (Figure 6).





Figure 6. The bacterial growth in exposure to A. annua nanoemulsion EO

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# **3.5.Anti-biofilm effects**

As shown in **Figure 7**, biofilms in the control group all bacteria were formed (OD > 0.083). However, after treatment with both nanoemulsions, no biofilm (OD < 0.083) was formed by all examined bacteria, i.e., *S. aureus, K. pneumonia, E. coli*, and *P. aeruginosa*.





Figure 7. The anti-biofilm effects of (mean OD value) samples

#### **4. Discussion**

۲٦۸ In our study, nanoemulsions of C. aurantium and A. annua EOs were prepared using spontaneous emulsification. Meanwhile, their antimicrobial and antibiofilm against selected 229 bacterial pathogens were investigated. Some previous studies have investigated antimicrobial ۲۷. and insecticide properties of C. aurantium and A. annua extracts and EOs against different 177 777 microorganisms, but our study has been carried out mostly on their nanoemulsion forms (22). 777 For example, the biological activity and antibiofilm molecular profile of *C. aurantium* EO were ۲۷٤ investigated by Kačániová et al., that concluded that C. aurantium EO had potent antibacterial 2007 activity against Stenotrophomonas maltophilia and Bacillus subtilis followed by Penicillium 272 crustosum (15). In addition, EOs extracted from the peel of C. aurantium, as noted by Madhuri et 777 al., can be used against infectious agents such as K. pneumoniae and Bacillus cereus (23). In ۲۷۸ *vitro* studies by Marinas et al. also showed that A. annua EO has selective antipathogenic activity ۲۷۹ on Gram-positive and Gram-negative bacterial strains (24). Its antibacterial and anti-biofilm ۲٨۰ activities have been also unveiled (25, 26). In addition, Mariadosset et al. fabricated the selenium

۲۸۱ nanoparticles using A. annua (AaSeNPs) of 109.2 nm in size and the characterized AaSeNPs ۲۸۲ indicated an antibacterial activity against multidrug-resistant pathogens such as S. aureus, B. ۲۸۳ subtilis, Proteus mirabilis, and E. coli (27). As well as the results obtained by Das et al. showed ۲۸٤ stronger antimicrobial activity of Pickering nanoemulsion of the Artemisia essential oil (26). In our study, antimicrobial and antibiofilm effects of nanoemulsions of C. aurantium and A. annua ۲۸٥ ۲۸٦ EOs against selected bacterial pathogens were confirmed by MIC and MBC tests with the ۲۸۷ highest activity against S. aureus. Exposure of pathogens with nanoemulsions resulted in a significant reduction in the biofilm formation by all examined bacteria i.e., S. aureus, K. ۲۸۸ ۲۸۹ pneumonia, E. coli, and P. aeruginosa. This study firstly investigated the chemical composition ۲٩. of C. aurantium and A. annua EO. Their nanoemulsion dosage forms were then prepared. The 291 antimicrobial and antibiofilm activities of nanomulsions were unraveled. Both nanoemulsions 292 inhibited the growth of S. aureus. Interestingly, no biofilms of S. aureus, K. pneumonia, E. coli, and P. aeruginosa were observed after treatment with these nanoemulsions. Moreover, A. annua ۲۹۳ 295 nanoemulsion inferred potent antioxidant properties. Limitations of this study mostly included low number of bacterial pathogens, lack of in silico assessment of binding of bioactive 290 297 compounds to bacterial targets and lack of in vivo study.

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#### **Y99** Authors' contributions

r..MO and AG conceptualized and designed the study. AG performed antibacterial tests andr..revised the MS. HA drafted MS in cooperation with MO. MAH and BM preparedr..nanoemulsions. EZ interpreted ATR-FTIR. MZ performed the DPPH assay. MO designed the

- $r \cdot r$  study, supervised the project, and drafted the MS. All authors contributed to drafting MS and
- $r \cdot \epsilon$  approved the final version.

# *v.o* Conflict of interest

- **٣.٦**None to declare.
- $\gamma \cdot \gamma$  Data availability
- r. A The data used to support the findings of this study are available from the corresponding author
- ۳۰۹ upon request.

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# **Ethics**

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